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## Chiral Amino Alcohol-Mediated Asymmetric Conjugate Addition of Arylalkynes to Nitroolefins

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## **ABSTRACT**

The asymmetric reaction of nitroolefins with arylalkynes was mediated by dimethylzinc (or diethylzinc) and (1R,2R)-2-(dimethylamino)-1,2-diphenylethanol in toluene to provide the corresponding conjugate alkynylation products with high enantiomeric excess of up to 99% in good yields. The presence of 0.03 equiv of galvinoxyl improved the reaction yield.

The asymmetric addition of alkynes to an electron-deficient double bond has been the recent focus of asymmetric reactions.<sup>1</sup> Aldehydes,<sup>2–5</sup> ketones,<sup>6–8</sup> and imines<sup>9–12</sup> are electrophiles that have been well documented by many

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- groups. On the other hand,  $\alpha,\beta$ -unsaturated carbonyl compounds or their congeners are acceptors that have seldom accomplished high enantioselectivity. The catalytic asymmetric conjugate alkynylation of enones recently re-
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ported by Corey and Chong is the exception.<sup>16</sup> We describe herein that arylalkynes undergo conjugate addition reaction with nitroolefins to provide the corresponding alkynylation adducts with high enantioselectivities of up to 99% ee.

Our approach began with the application of the chiral ligand-controlled asymmetric conjugate addition reaction of aryllithiums with nitroolefins. The reaction of lithiated trityl ether of propargyl alcohol **2a** with nitroolefin **1a** in the presence of a tridentate chiral amino ether ligand **3a** in toluene, however, gave the addition product **4a** with a marginal ee of 7% in 96% yield. Reaction of heteroaggregated lithium acetylide of **2a** with a chiral amino alcohol **3b**, the starting material for the preparation of **3a**, <sup>17b</sup> was not successful, giving **4a** with 4% ee in 92% yield.

Promising results were obtained by changing the metal from lithium to zinc (Table 1). The reaction of 1.5 equiv of

**Table 1.** Amino Alcohol **3b**-Mediated Asymmetric Conjugate Addition of **2b** to **1a** Giving **4b** 

entry	<b>2b</b> (equiv)	<b>3b</b> (equiv)	$\begin{array}{c} Me_2Zn\\ (equiv) \end{array}$	temp (°C)	time (h)	yield (%) <sup>a</sup>	ee (%) <sup>b</sup>
1	1.5	1.5	1.5	50	24	39	95
$2^c$	1.5	1.5	1.5	rt	120	52	96
3	3	3	3	50	10	66	95
$4^d$	3	3	3	rt	30	83	97
$5^{d,e}$	3	3	3	rt	30	74	97
6	3	0	3	50	15	< 5	

 $^a$  Combined yield.  $^b$  Ee (%) of trans-4b.  $^c$  DMPEG (0.1 equiv) was used.  $^d$  Galvinoxyl (0.03 equiv) was used.  $^e$  Diethylzinc was used in place of dimethylzinc.

phenylacetylene **2b** with nitroolefin **1a** was conducted in the presence of 1.5 equiv of **3b** and 1.5 equiv of dimethylzinc in toluene at 50 °C for 24 h to give a 64:36 mixture of *trans*-and *cis*-**4b** with 95% ee and 94% ee, respectively, in 39%

yield (Table 1, entry 1). The enantiomeric excess was determined by chiral stationary phase HPLC (Daicel Chiralcel OD-H, hexane/*i*-PrOH = 100/1; See Supporting Information). The yield was slightly improved by the addition of 0.1 equiv of poly(ethylene glycol) dimethyl ether (DMPEG)<sup>19</sup> at room temperature for 120 h to give **4b** with 96% ee in 52% yield (entry 2). Screening of solvents (DMSO, THF, acetonitrile, CH<sub>2</sub>Cl<sub>2</sub>) was not fruitful, giving **4b** with 96–97% ee in 19–43% yield.

Improvement of the yield was achieved by using 3 equive ach of **2b**, **3b**, and dimethylzinc in toluene at 50 °C for 10 h to give **4b** with 95% ee in 66% yield (entry 3). Much more improvement was obtained upon addition of 0.03 equivof galvinoxyl to scavenge radicals, <sup>20</sup> giving, after 30 h at room temperature, a 73:27 mixture of *trans*- and *cis*-**4b** with 97 and 96% ee, respectively, in 83% yield (entry 4). Dimethylzinc was replaceable with diethylzinc to give **4b** in comparable enantioselectivity and yield (entry 5). It is also important to note that in the absence of chiral amino alcohol **3b**, the reaction at 50 °C for 15 h gave **4b** in less than 5% yield (entry 6).

It is important to note that saturated ammonium chloride quench of the reaction gave a trans major 73:27 mixture of **4b**, while acetic acid quench favored *cis*-**4b** in a ratio of 81:19. Isomerization of pure *cis*-**4b** of 94% ee was possible by treatment with triethylamine in acetonitrile at room temperature for 24 h to give **4b** as a trans major 71:29 mixture in 95% yield without any racemization.

The chirality controller of amino alcohol **3b** is recoverable in 95% yield by collecting its crystalline hydrochloride after 10% hydrochloric acid treatment of the quenched reaction mixture and is reusable in the asymmetric reaction.

Linear nitroolefins  $\mathbf{1b-d}$  (R = n-Bu, i-Pr, t-Bu) were also reacted with phenylacetylene  $\mathbf{2b}$  under the established conditions as above to give the corresponding conjugate alkynylation products  $\mathbf{5b-d}$  with 98, 96, and 98% ee, respectively, in moderate to good yields (Table 2). The moderate yields in entries 1 and 2 are due to concomitant deprotonation and isomerization of double bond of  $\mathbf{1b}$  and  $\mathbf{1c}$ .

**Table 2.** Amino Alcohol **3b**-Mediated Asymmetric Conjugate Addition of **2b** to Linear Olefins **1b-d** Giving **5** 

entry	1	R	$temp\ (^{\circ}C)$	time (h)	yield (%)	ee (%)
$1^a$	1b	n-Bu	rt	24	32	98
2	1c	$i ext{-}\mathrm{Pr}$	rt	72	44	96
3	1d	t-Bu	50	48	88	98

<sup>&</sup>lt;sup>a</sup> Diethylzinc was used in place of dimethylzinc.

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Arylacetylenes are the choice of nucleophiles in the present asymmetric conjugate addition reaction. The reaction of 4-tolyl-, 4-fluorophenyl, 1-naphthyl-, and 3-pyridylacetylenes **2c**-**f** reacted with nitroolefin **1a** to give **4c**-**f** with 96-99% ee in good to moderate yields (Table 3).

**Table 3.** Amino Alcohol **3b**-Mediated Asymmetric Conjugate Addition of **2b**-**f** to **1a** Giving **4b**-**f** 

entry	2	Ar	temp (°C)	time (h)	yield $(\%)^a$	ee (%) $^b$
1	<b>2</b> b	Ph	rt	30	83	97
2	2c	Tolyl	$\mathbf{rt}$	24	85	$98^{c}$
3	2d	$4\text{-FC}_6H_5$	50	20	73	98
4	2e	1-Naph	$\mathbf{rt}$	48	82	99
5	2f	3-Py	50	48	30	97

 $^a$  Combined yield. The ratio of trans:cis ranges from 73:27 to 21:79 (See Supporting Information).  $^b$  Ee (%) of *trans-4* unless otherwise mentioned.  $^c$  Ee (%) of cis-4c.

The absolute configuration of **4b** was determined to be (R)- by the conversion to (+)-**7** with the established absolute configuration (Scheme 1).<sup>21</sup> The absolute configuration of other alkynylation products was tentatively assigned as shown by the reaction analogy.

**Scheme 1.** Determination of the Absolute Configuration of **4b** by Conversion to the Ketone **7** with the Established Absolute Configuration

The sense of asymmetric induction is predictable by the model **8** originally proposed by Noyori and Oguni<sup>22</sup> in the

Figure 1. Model 8 predicting the sense of asymmetric induction.

chiral amino alcohol-mediated catalytic asymmetric alkylation of aldehydes with dialkylzinc reagents (Figure 1). A zinc metal in the bimetallic complex  $\bf 8$  serves as a Lewis acid that coordinates to the oxygen of a nitro group, and then an alkynyl group attacks the double bond to complete the reaction, giving ( $\it R$ )-4 $\it b$ . The ( $\it R$ )-aboslute configuration predicted in  $\it 8$  is identical with that observed.

Not only nitroolefin 1 but also enone is the substrate of the asymmetic conjugate alkynylation. The reaction of 2b with chalcone 9 proceeded smoothly at 70 °C for 96 h to give the conjugate alkynylation product 10<sup>16a</sup> with 82% ee in 92% yield (Scheme 2). Thus the present conditions have

the possibility of being widely applicable to other types of activated olefins.

In conclusion, we have developed the highly enantioselective conjugate addition reaction of arylalkyne with nitroolefin by the mediation of chiral amino alcohol and diorganozinc. However, development of a catalytic protocol and the use of a nonarylalkyne are the targets of the next stage.

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**Supporting Information Available:** Reaction procedure and the characterization data of new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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